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## Key indicators

Single-crystal X-ray study
$T=100 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.002 \AA$
$R$ factor $=0.031$
$w R$ factor $=0.078$
Data-to-parameter ratio $=17.4$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## 1-Benzyl-3-[2-(1-piperidinio)ethyl]benzimidazolium dichloride monohydrate

The title compound, $\mathrm{C}_{21} \mathrm{H}_{27} \mathrm{~N}_{3}^{2+} \cdot 2 \mathrm{Cl}^{-} \cdot \mathrm{H}_{2} \mathrm{O}$, was synthesized from 1-benzylbenzimidazole and 2-chloroethylpiperidine hydrochloride in dimethylformamide. In the cation, the benzimidazole ring is connected to the piperidine ring by an ethylene group. The crystal structure is stabilized by $\mathrm{O}-$ $\mathrm{H} \cdots \mathrm{Cl}, \mathrm{N}-\mathrm{H} \cdots \mathrm{Cl}, \mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ and $\mathrm{C}-\mathrm{H} \cdots \mathrm{Cl}$ hydrogenbonding interactions.

## Comment

Heterocyclic compounds generally exhibit versatile pharmacological activities, such as antitumour, diuretic, fungicidal, bactericidal, antihelmintic, antiallergic, vasodilator, antihistaminic and local analgesic. We have reported the synthesis and antimicrobial activities of many benzimidazole derivatives (Küçükbay et al., 2003, 2004) and elucidated the crystal structures of some benzimidazole derivatives having piperidine or morpholine groups (Türktekin et al., 2004; Akkurt, Türktekin et al., 2004; Akkurt, Öztürk et al., 2004; Akkurt et al., 2005). We now report the synthesis and crystal structure of a biologically interesting piperidine-substituted benzimidazole compound, (I).

(I)

The molecular structure of (I) is illustrated in Fig. 1, and selected bond distances and angles are given in Table 1. All the geometric parameters of (I) agree with the results obtained in our previous studies of related heterocycles (Akkurt et al., 2005; Türktekin et al., 2004). The benzimidazole ring is essentially planar, with maximum deviations of 0.012 (1) $\AA$ for atom N1 and -0.012 (1) $\AA$ for atom C6. The dihedral angle between the planes of the phenyl and benzimidazole ring systems is $72.83(6)^{\circ}$. The piperidine ring has a chair conformation [the puckering parameters (Cremer \& Pople, 1975) are $Q_{\mathrm{T}}=0.5701(18) \AA, \theta=177.12(18)^{\circ}$ and $\left.\varphi=195(4)^{\circ}\right]$.

The crystal structure of (I) is stabilized by $\mathrm{O}-\mathrm{H} \cdots \mathrm{Cl}, \mathrm{N}-$ $\mathrm{H} \cdots \mathrm{Cl}, \mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ and $\mathrm{C}-\mathrm{H} \cdots \mathrm{Cl}$ hydrogen-bonding interactions. Details are given in Table 2 and the hydrogen-bonding involving the Cl 1 anion and the water molecule of crystallization, O1, is illustrated in Fig. 2.

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An ORTEP-3 (Farrugia, 1997) view of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the $50 \%$ probability level.

## Experimental

Compound (I) was synthesized by heating, on a water bath for 3 h , a mixture of 1-benzylbenzimidazole ( $2.00 \mathrm{~g}, 9.6 \mathrm{mmol}$ ) and 2-chloroethylpiperidine hydrochloride $(1.8 \mathrm{~g}, 9.6 \mathrm{mmol})$ in dimethylformamide ( 10 ml ). The volatiles were then removed under vacuum and the crude solid obtained was crystallized from an $\mathrm{EtOH} / \mathrm{Et}_{2} \mathrm{O}$ (3:1) mixture. Colourless plate-like crystals were obtained (yield 3 g , $79 \%$; m.p. $469-470 \mathrm{~K}) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}\right): 81.4-1.8(q$, piperidine, 6 H$)$, 3.2 ( $q$, piperidine, 4 H ), 3.6 ( $t, \mathrm{CH}_{2} \mathrm{CH}_{2}$-piperidine, 2 H ), 4.8 ( $t$, $\mathrm{CH}_{2} \mathrm{CH}_{2}$-piperidine, 2 H ), 4.01 ( $t$, ring methylene, 4 H ), 4.13 ( $q$, $\left.\mathrm{CH}_{2} \mathrm{CH}_{3}, 2 \mathrm{H}\right), 4.56\left(q,-\mathrm{CH}_{2} \mathrm{CH}_{3}, 2 \mathrm{H}\right), 5.5\left(s, \mathrm{CH}_{2}-\mathrm{Ph}, 2 \mathrm{H}\right), 7.3(s$, $\mathrm{Ar}-\mathrm{H}, 4 \mathrm{H}), 9.3$ ( $s$, benzimidazole-C2-H, 1 H ). Analysis calculated for $\mathrm{C}_{21} \mathrm{H}_{29} \mathrm{Cl}_{2} \mathrm{~N}_{3} \mathrm{O}$ : C 61.46, H 7.07, $\mathrm{N} 10.24 \%$; found: C 63.02, H 6.71, N 10.70\%.

## Crystal data

$\mathrm{C}_{21} \mathrm{H}_{27} \mathrm{~N}_{3}^{2+} \cdot 2 \mathrm{Cl}^{-} \cdot \mathrm{H}_{2} \mathrm{O}$
$M_{r}=410.37$
Monoclinic, $P 2_{1} / n$
$a=17.1422$ (12) $\AA$
$b=6.9643$ (3) A
$c=18.2627$ (13) $\AA$
$\beta=106.961$ (5) ${ }^{\circ}$
$V=2085.4$ (2) $\AA^{3}$
$Z=4$

## Data collection

Stoe IPDS-II diffractometer $\omega$ scans
Absorption correction: integration ( $X$-RED32; Stoe \& Cie, 2002) $T_{\text {min }}=0.784, T_{\text {max }}=0.943$
17638 measured reflections
4535 independent reflections

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.031$
$w R\left(F^{2}\right)=0.078$
$S=1.04$
4535 reflections
260 parameters
H atoms treated by a mixture of independent and constrained refinement
$D_{x}=1.307 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
Cell parameters from 17638 reflections
$\theta=1.9-27.2^{\circ}$
$\mu=0.33 \mathrm{~mm}^{-1}$
$T=100 \mathrm{~K}$
Plate, colourless
$0.78 \times 0.49 \times 0.18 \mathrm{~mm}$

3875 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.032$
$\theta_{\text {max }}=27.0^{\circ}$
$h=-21 \rightarrow 21$
$k=-8 \rightarrow 8$
$l=-23 \rightarrow 22$

$$
\begin{gathered}
w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.0383 P)^{2}\right. \\
\quad+0.7441 P] \\
\text { where } P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
(\Delta / \sigma)_{\max }=0.001 \\
\Delta \rho_{\max }=0.43 \mathrm{e}^{-3} \AA^{-3} \\
\Delta \rho_{\min }=-0.18 \mathrm{e}^{-3}
\end{gathered}
$$



Figure 2
The crystal packing of (I), viewed along the $b$ axis. Dashed lines indicate $\mathrm{O}-\mathrm{H} \cdots \mathrm{Cl}$ hydrogen-bonding contacts (details are give in Table 2).

Table 1
Selected geometric parameters ( $\left(\mathrm{A},{ }^{\circ}\right)$.

| N1-C1 | $1.3947(16)$ | $\mathrm{N} 2-\mathrm{C} 15$ | $1.4637(16)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{N} 1-\mathrm{C} 14$ | $1.3279(16)$ | $\mathrm{N} 3-\mathrm{C} 17$ | $1.5004(18)$ |
| $\mathrm{N} 1-\mathrm{C} 7$ | $1.4747(18)$ | $\mathrm{N} 3-\mathrm{C} 16$ | $1.4834(17)$ |
| N2-C14 | $1.3342(18)$ | $\mathrm{N} 3-\mathrm{C} 21$ | $1.4974(19)$ |
| N2-C6 | $1.3973(16)$ |  |  |
| C1-N1-C7 | $125.86(11)$ | $\mathrm{N} 1-\mathrm{C} 1-\mathrm{C} 6$ | $106.61(10)$ |
| C1-N1-C14 | $108.36(11)$ | $\mathrm{N} 2-\mathrm{C} 6-\mathrm{C} 1$ | $106.42(11)$ |
| C7-N1-C14 | $125.76(11)$ | $\mathrm{N} 2-\mathrm{C} 6-\mathrm{C} 5$ | $131.60(11)$ |
| C6-N2-C14 | $108.17(10)$ | $\mathrm{N} 1-\mathrm{C} 7-\mathrm{C} 8$ | $112.37(11)$ |
| C6-N2-C15 | $126.09(11)$ | $\mathrm{N} 1-\mathrm{C} 14-\mathrm{N} 2$ | $110.43(11)$ |
| C14-N2-C15 | $125.74(11)$ | $\mathrm{N} 2-\mathrm{C} 15-\mathrm{C} 16$ | $109.17(10)$ |
| C16-N3-C21 | $109.32(11)$ | $\mathrm{N} 3-\mathrm{C} 16-\mathrm{C} 15$ | $112.05(11)$ |
| C17-N3-C21 | $111.10(11)$ | $\mathrm{N} 3-\mathrm{C} 17-\mathrm{C} 18$ | $110.35(13)$ |
| C16-N3-C17 | $112.11(11)$ | $\mathrm{N} 3-\mathrm{C} 21-\mathrm{C} 20$ | $110.29(13)$ |
| N1-C1-C2 | $131.10(12)$ |  |  |

Table 2
Hydrogen-bond geometry ( $\AA{ }^{\circ}{ }^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{~N} 3-\mathrm{H} 3 A \cdots \mathrm{Cl} 2$ | $0.92(2)$ | $2.10(2)$ | $3.0066(13)$ | $171(2)$ |
| $\mathrm{O} 1-\mathrm{H} 22 \cdots \mathrm{Cl} 1^{\mathrm{i}}$ | $0.84(2)$ | $2.35(2)$ | $3.1919(12)$ | $176(2)$ |
| $\mathrm{O} 1-\mathrm{H} 23 \cdots \mathrm{Cl} 1^{\text {ii }}$ | $0.85(2)$ | $2.33(2)$ | $3.1722(12)$ | $175(2)$ |
| $\mathrm{C} 2-\mathrm{H} 2 \cdots \mathrm{Cl} 2{ }^{\text {iii }}$ | 0.93 | 2.81 | $3.6549(14)$ | 152 |
| $\mathrm{C} 7-\mathrm{H} 7 A \cdots \mathrm{Cl} 1$ | 0.97 | 2.80 | $3.6930(15)$ | 153 |
| $\mathrm{C} 9-\mathrm{H} 9 \cdots \mathrm{Cl} 2^{\text {iii }}$ | 0.93 | 2.75 | $3.6243(15)$ | 158 |
| $\mathrm{C} 15-\mathrm{H} 15 A \cdots \mathrm{O} 1^{\mathrm{ii}}$ | 0.97 | 2.57 | $3.2734(18)$ | 129 |
| $\mathrm{C} 15-\mathrm{H} 15 B \cdots \mathrm{Cl} 2$ | 0.97 | 2.79 | $3.5749(15)$ | 139 |
| Symmetry codes: (i) | $x, y-1, z ;$ (ii) | $-x+1,-y+1,-z+1 ;$ (iii) | $-x+\frac{3}{2},+y+\frac{1}{2}$, |  |
| $-z+\frac{3}{2}$. |  |  |  |  |

## organic papers

All H atoms were found in difference Fourier maps. The water molecule H atoms and the N -bound H atom were refined isotropically. The other H atoms were refined with a riding model, with $\mathrm{C}-\mathrm{H}=0.93-0.97 \AA$, and with $U_{\text {iso }}$ constrained to be $1.2 U_{\text {eq }}$ of the carrier atom.

Data collection: $X$-AREA (Stoe \& Cie, 2002); cell refinement: $X$-AREA; data reduction: X-RED32 (Stoe \& Cie, 2002); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999).

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